ORAL MICROBIOLOGY AND IMMUNOLOGY

# Characterization of ∟-cysteine desulfhydrase from *Prevotella intermedia*

Yano T, Fukamachi H, Yamamoto M, Igarashi T. Characterization of L-cysteine desulfhydrase from Prevotella intermedia.

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**Introduction:** Hydrogen sulfide is responsible for lysis of red blood cells and is a major compound for oral malodor. To clarify the production mechanism of hydrogen sulfide in *Prevotella intermedia*, we found an L-cysteine desulfhydrase gene (*lcs*) homologue on the genome database of *P. intermedia* ATCC25611 and characterized its gene product. **Methods:** The *lcs* gene homologue cloned into pGEX6p-1 vector was expressed in *Escherichia coli* and purified. Lcs activity was assayed by detection of the reaction products (hydrogen sulfide and pyruvate) or its derivatives from L-cysteine. Site-directed mutagenesis was used to convert an amino acid of the Lcs molecule.

**Results:** The purified *lcs* gene product catalysed the degradation of L-cysteine to pyruvate, ammonia, and hydrogen sulfide, indicating that the protein is L-cysteine desulfhydrase. The enzyme required pyridoxal 5'-phosphate as a cofactor, and it was highly active at pH 7.0 and completely inhibited by ZnCl<sub>2</sub>. The  $K_{\rm m}$  and  $V_{\rm max}$  of the enzyme were 0.7 mM and 4.2  $\mu$ mol/min/mg, respectively. Replacement of Tyr-59, Tyr-118, Asp-198, and Lys-233 with any of the amino acids resulted in the complete disappearance of Lcs activity, implying that these amino acids are essential for enzyme activity. In addition, hydrogen sulfide produced by this enzyme lysed sheep red blood cells and modified hemoglobin.

**Conclusion:** These results show the enzymatic properties of L-cysteine desulfhydrase from *P. intermedia* ATCC25611 and also suggest that the Lcs enzyme, which produces hydrogen sulfide from L-cysteine, is closely associated with the pathogenesis of *P. intermedia*.

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*Prevotella intermedia*, a gram-negative, black-pigmented anaerobic rod, is frequently isolated from periodontal lesions associated with various forms of periodontal diseases such as chronic periodontitis, pregnancy gingivitis, and acute necrotizing ulcerative gingivitis (17, 20, 26, 33, 45). This bacterium produces a variety of virulence factors including adhesins, fimbriae, hemagglutinin, hydrolytic enzymes, and hemolysin (21, 31, 39, 42, 43) and may play a role in the initiation and development of periodontal disease (26, 33, 41, 45). Like other periodontal pathogens, *P. intermedia* requires iron for growth; however, it lacks the ability to synthesize heme, an important growth factor for the bacterium (18, 28). Therefore, sufficient mechanisms to acquire iron are needed to survive and proliferate in the host. Hemo-globin or hemin derived from the hemolytic degradation of red blood cells is a major source of exogenous iron for the nutrient of bacterial growth. There are several reports that *P. intermedia* has hemolytic and hemagglutinating activities (4, 39, 42). Although a hemolysin activated by reduc-

tants from *P. intermedia* was partially purified and characterized (43) and a gene encoding multiple hemolytic domains from *P. intermedia* was cloned (3), the hemolytic substances and the mechanism of hemolysis by *P. intermedia* are still unknown.

*P. intermedia* is frequently detected from patients with not only periodontal diseases but also oral malodor (27, 44). Previous studies have suggested a high correlation between periodontal diseases and oral malodor (12, 36, 46). The major compounds contributing to oral malodor are volatile sulfur compounds (VSC), including hydrogen sulfide, methyl mercaptan, and dimethyl sulfide (25), and these VSC have been found at relatively high levels in periodontal pockets (5, 11, 40). Several studies reveal that of these VSC, hydrogen sulfide induces lysis of red blood cells (RBCs) (29, 37, 47) and is highly toxic for mammalian cells (2). It is reported that genera such as Fusobacterium, Prevotella, Porphyromonas, Selenomonas, Veillonella, Peptostreptococcus, and Eubacterium are the predominant producers of hydrogen sulfide (40). Genetic and biochemical analyses have shown that hydrogen sulfide is produced by a bacterial enzyme. Until now, this type of enzyme has been identified, cloned, and characterized in oral bacteria such as Treponema denticola (9), Fusobacterium nucleatum (15), and Streptococcus anginosus (47). These studies demonstrate that hydrogen sulfide is produced from L-cysteine by L-cysteine desulfhydrase, which catalyses the  $\alpha,\beta$ -elimination of L-cysteine to produce hydrogen sulfide, pyruvate, and ammonia. As P. intermedia is a predominant producer of hydrogen sulfide (40), the presence of the L-cysteine desulfhydrase-like enzyme in P. intermedia would be expected. Therefore, finding and analysing the enzyme participating in the production of hydrogen sulfide would promote the understanding of the acquisition mechanism of iron (hemoglobin or hemin) in P. intermedia.

In this study, we cloned the gene homologue encoding L-cysteine desul-fhydrase in *P. intermedia* ATCC25611 and characterized its gene product. In addition, the hemolytic activity and hemo-globin modification by this enzyme were examined.

# Materials and methods Bacterial strains and culture conditions

*P. intermedia* ATCC25611 was grown anaerobically (10% CO<sub>2</sub>, 10% H<sub>2</sub>, 80% N<sub>2</sub>) at 37°C in brain–heart infusion broth (Difco Laboratories, Detroit, MI) with hemin (5  $\mu$ g/ml) and menadione (50  $\mu$ g/ml). *Escherichia coli* BL21 was used as a plasmid host and cultured aerobically at 37°C in Luria–Bertani broth (Invitrogen, Carlsbad, CA).

# **DNA** manipulation

Chromosomal DNA of *P. intermedia* ATCC25611 was extracted with a GenElute Bacterial genomic DNA kit (Sigma-Aldrich, St. Louis, MO). Plasmid DNA was extracted with a Wizard miniprep purification kit (Promega, Madison, WI). Ligation of DNA was performed with a DNA Ligation Kit version 2 (TakaRa Bio, Shiga, Japan) and site-directed mutagenesis of the DNA fragments was carried out with a QuikChange site-directed mutagenesis kit (Stratagene, La Jolla, CA) as recommended by the supplier (22).

# DNA sequence of L-cysteine desulfhydrase gene homologue of *P. intermedia* ATCC25611

The DNA fragment [1547 base pairs (bp) long] containing the L-cysteine desulfhydrase gene (lcs) homologue from P. intermedia ATCC25611 was amplified with the primer pair designed on the basis of the PI0305 sequence, which encodes putative L-cysteine desulfhydrase, on the P. intermedia strain 17 genome DNA database (http://www.oralgen.lanl.gov/oralgen/ bacteria/pintnew/). The primer sequences were as follows: forward, 5'-GACC GTGCATTCGAAGACAG-3' and reverse. 5'-CTAAGTATTTCTGAATTACAA-3'. Long and accurate polymerase chain reaction (LA-PCR) with LA Taq DNA polymerase (TakaRa Bio) was carried out to amplify the gene and Shuttle PCR was used as a condition of amplification, as recommended by the supplier (TakaRa Bio) (23). The amplified fragment (1547 bp) containing the *lcs* homologue (1170 bp) was cloned into pT7 Blue T-vector (Novagen, Madison, WI) and subcloned for DNA sequencing.

# Cloning, expression, and purification of recombinant Lcs

The lcs gene homologue from P. intermedia ATCC25611 was amplified by LA-PCR with LA Taq DNA polymerase. PCR primers were designed on the basis of the nucleotide sequence of the lcs homologue from P. intermedia ATCC25611 determined in this study. The primer sequences were as follows: forward, 5'-GGG GAATTCATGGGAAAATATAATTTCGA CGAGC-3' and reverse, 5'-GGGGTCGAC TTCACAGGCTTTCAACACGTTTACG-3'. The underline in each primer indicates the EcoRI and SalI restriction sites, respectively. Shuttle PCR was used as a condition of amplification as recommended by the supplier (TakaRa Bio) (23). Briefly, the reaction mixture (50  $\mu$ l) was denatured at 94°C for 1 min followed by 28 cycles of 20 s at 98°C for denaturation and 2 min at 68°C for annealing and extension. The amplified fragment (1170 bp) was cloned into a pGEX-6P-1

vector (GE Healthcare UK, Little Chalfont, UK) and the resulting plasmid was designated as pCDS1. The pCDS1 was transformed into *E. coli* BL21 cells and expressed. The recombinant Lcs protein was purified by the GST purification system according to the manufacturers' instructions (GE Healthcare).

#### Assay of ∟-cysteine desulfhydrase activity

L-Cysteine desulfhydrase activity was detected using three different assays.

# Assay I: formation of methylene blue

The assay was carried out in a 1-ml reaction mixture containing 100 mM Tris-HCl (pH 7.0), 2.5 mM dithioerythritol, 10 µM pyridoxal-5'-phosphate (PLP), 1 mM L-cysteine, and enzyme. After incubation at 37°C, the reaction was terminated by the addition of 0.1 ml of 20 mM N'.N'dimethyl-p-phenylenediamine dihvdrochloride dissolved in 7.2 M HCl and 0.1 ml of 30 mM FeCl<sub>3</sub> dissolved in 1.2 M HCl. After 20 min of incubation at room temperature, the formation of methylene blue was determined at an optical density at 670 nm (OD<sub>670</sub>) using a molar extinction coefficient of  $28.5 \times 10^6$  (15).

# Assay II: formation of black precipitate from bismuth

The sample was suspended in a visualizing solution containing 100 mM triethanolamine-HCl (pH 7.0), 0.5 mM bismuth trichloride, 1% Triton X-100, 10 mM ethylenediaminetetraacetic acid, 10 µM PLP, and 5 mM L-cysteine. The bismuth in this solution reacts with sulfide, forming a black precipitate. This method was used to visualize the hydrogen sulfide-producing activity after non-denaturing polyacrylamide gel electrophoresis (native-PAGE) as described previously (47). After electrophoresis of the enzyme, the gel was soaked in the visualizing solution at 37°C. A black precipitate formed at the position of the enzyme producing hydrogen sulfide in the gel.

# Assay III: formation of pyruvate

The assay was carried out with 200  $\mu$ l of 100 mM Tris–HCl (pH 7.0) containing 10  $\mu$ M PLP and 1 mM L-cysteine. The enzyme was added to start the reaction. After incubation for 20 min at 37°C, the reaction was terminated by adding 100  $\mu$ l of 4.5% trichloroacetic acid. The suspension was centrifuged, and 250  $\mu$ l of the The kinetic parameters ( $K_{\rm m}$  and  $V_{\rm max}$ ) were computed from a Lineweaver–Burk transformation ( $V^{-1}$  vs.  $S^{-1}$ ) of the Michaelis–Menten equation, where V is the formation rate of hydrogen sulfide ( $\mu$ mol/ min/mg) and S is the concentration (M) of L-cysteine.

# Preparation of Lcs mutants

pCDS1 containing the lcs gene was used for the preparation of Lcs mutants. Replacement of amino acids in the Lcs protein was carried out using a Quik-Change site-directed mutagenesis kit (Stratagene) (22). The nucleotide sequences of the mutagenic PCR primers designed to replace a wild-type amino acid (Y59, Y118, D198, K233, or R365) with the other amino acids are as follows: 5'-CACGGTGTCTTCGGCGC Y59A, TGCCATTGTGGGGCGAC-3'; Y118A, 5'-ATACAGACACCAGTCGCCAATTG TTTCTATTCT-3'; D198N, 5'-GTGAAG GTGATAAGCAACGAAATTCATTGCG AG-3'; K233N, 5'-TTCAATTCGCCTAC CAATAACTTCAACATCGCA-3'; R365G, 5'-GGCGAGGGCTATATCGGAATCAAC CTTGCTTGC-3', and their complementary oligonucleotides primers, respectively. The mutated bases are underlined on each primer. The mutations generated in the lcs gene were confirmed by nucleotide sequencing. The resultant mutated lcs gene was expressed in E. coli BL21 cells and purified using the GST purification system as described above (23).

# Assay of hemolysis and hemoglobin modification

Hemolysis assay was performed as described by Chu et al. with some modifications (7). Hemoglobin and modifiedhemoglobin in sheep erythrocytes (RBCs) were measured as described previously (47) with some modifications. Briefly, RBCs were washed three times with phosphate-buffered saline to remove the soluble hemoglobin before the RBCs were used. The reaction mixture consisted of 120 mM NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub>, 5 mM KH<sub>2</sub>PO<sub>4</sub> (pH 7.5), 10 μM PLP, 4 μg/ml Lcs and 4% (volume/volume) RBCs with 1 mM L-cysteine. The mixture was incubated at 37°C with slow rotation and was withdrawn at 0, 3, 6, 9, and 12 h for

measuring the hemolysis of RBCs and the hemoglobin and modified-hemoglobin in RBCs. Hemolysis was estimated by measuring the unbroken RBCs in the reaction mixture at OD<sub>405</sub>. The percent hemolysis was calculated as described previously (7). The hemoglobin-modification assay was carried out as follows: the reaction mixture withdrawn at the time indicated above was centrifuged to collect the RBCs. The RBCs were washed several times, suspended in phosphate-buffered saline and then lysed by the addition of distilled water. Hemoglobin and modifiedhemoglobin in the lysate were measured at OD<sub>540</sub> and OD<sub>620</sub>, respectively, to observe the changes in the amounts of hemoglobin and modified-hemoglobin in RBCs (47).

#### Nucleotide sequence accession number

The nucleotide sequence of *lcs* gene from *P. intermedia* ATCC25611 reported in this paper has been deposited in the DNA Data Bank of Japan nucleotide sequence databases with accession number AB481366.

#### Results

## Analysis of L-cysteine desulfhydrase gene from *P. intermedia* ATCC25611

To determine and analyse the L-cysteine desulfhydrase gene from *P. intermedia* ATCC25611, a homologue of the gene was searched on the genome database of *P. intermedia* strain 17 on the basis of the amino acid sequences of Hly (6), Lcd (47), MalY (50), and PatB (1), which were

MGKYNFDELVNRRGTNSVKWDEEKE----DGIIPLWVADMDFLAAPAIRRAVEERAKH MSKYNFQTAPNRLSHHTYKWKETET----DPQLLPAWIADMDFEVMPEVKQAIHDYAEQ Lcs Lcd -MIYDFTTKISRKNLGSLKWDLMYSQNPEVGNEVVPLSVADMEFKNPPELIEGLKKYLDE Hly Malv -- MFDFSKVVDRHGTWCTQWDYVADRFG-- TADLLPFTISDMDFATAPCIIEALNQRLMH ---MNFDKREERLGTQSVKWDKTGELFG--VTDALPMWVADMDFRAPEAITEALKERLDH PatB GVFGYAIV-GDSYYAAITNWFKRRHNWAIERDWIIYTTGVIPAISATIHALAMPGEVVI LVYGYTYA-SDELLQAVLDWEKSEHQYSFDKEDIVFVEGVVPAISIAIQAFTKEGEAVLI TVLGYTGP-TEEYKKTVKKWMKDRHQWDIETDWIINTAGVVPAVFNAVREFTKPGDGVII Lcs Led Hly  $\label{eq:gvfg} gvfg YsrwkndeflaaiahwfstQh Ytaidsqtvvygpsviymvselirqwsetgegvvigffg Yttp-dqktkdavcgwmqnrhgwkvnpesitfspgvvtalsmavqaftepgdqvvv$ Maly PatB 180 QTPVYNCFYSCIRNQGLRVLESPLKREDDTYVVDWDDFEAKCADKKTTLFLLCTPHNPAG Lcs N SPVYPPFARSVRINNRKLVSNSLKEENGLFQIDFEQLEKDIVENNVKLYLLCSPHNPGG ITPVYYPFFMAIKNQERKIIECELLEKDGYYTIDFEKLEKLSKDKNNKALLFCSPHNPVG Lcd Hly Maly A Y D A F Y K A I E G N Q R T V M P V A L E K Q A D G W F C D M G K L E A V L A K P E C K I M L L C S P Q N P T G PatB QPPVYTPFYHMVEKNGRHILHNPLLEKDGAYAIDFEDLETKLSDPSVTLFILCNPH<mark>NP</mark>SG 240 Z40 RVWSKDELARMNEICMRHDVKVISDEIHCELIMPGYVFTPFATVNADCQRNSVVFNSPTK RIWEREVLERIGHLCQKHQVILVSDEIHQDLTLFSHEHVSFNTISPDFKEFAIILSSATK RVWKKDELQKIKDIVLKSDLMLWSDEIHFDLIMPGYEHTVFQSIDEQLADKTITFTAPSK Lcs Lcd Hlv Maly KVWTCDELEIMADLCERHGVRVISDEI HMDMVWGEQPHIPWSNVAR---GDWALLTSGSK RSWSREDLLKLGELCLEHGVTVVSDEIHSDLMLYGHKHTPFASLSDDFADISVTCAAPSK PatB 300 NFNIAGLQIANIICADKEWYRRIDRAIN - IFEVCDVNPFGPLALEAAYNESEDWIDELMP TFNIAGTKNSYAVIENPSLRAQFQRQQL - ANNHHEVSSLGYIATETAYRYGRPWLVALKA TFNIAGMGMSNIIIKNPDIRERFTKSRD - ITSGMPFTTLGYKACEICYKECGKWLDGCIK SFNIPALTGAYGIIENSSSRDAYLSALKGRDGLSSPSVLALTAHIAAYQQGAPWLDALRI Lcs Lcd Hly MalY TFNIAGLOASAIIIPDRLKRAKFSASLQ-RNGLGGLNAFAVTAIEAAYSKGGPWLDELIT PatB Y TADNYALI, KDTFAKEVPNYEVI, KI, EGTYI, AWVDI RKSGI, TANAI, TEKI, I, REGKVOVNSG Les VLEENIKFAVDYFAKEAPRLKVMKPQGTYLIWLDFSDYGLTDDELFTLLHDQAKVILNRG Lcd VIDKNQRIVKDFFEVNHPEIKAPLIEGTYLQWIDFRALKMDHKAMEEFMIHKAQIFFDEG YLKDNLTYIADKMNAAFPELNWQIPQSTYLAWLDLRPLNIDDNALQKALIEQEKVAIMPG Hly MalY YIEKNMNEAEAFLSTELPKVKMMKPDASYLIWLDFSAYGLSDAELQQRMLKKGKVILEPG PatB Lcs VIYSKNDGEGYIRINLACPRATLOEGLKRIVNVLKACV-----SDYGKE-GELHARLNIATPKPLVEEICKRIVHCLPQ----Lcd YIFGDG-GIGFERINLAAPSSVIQESLERLNKALKDLKNRHLK Hly Maly YTYGEE - GRGEVRLNAGCPRSKLEKGVAGLINAIRAVR-----TKYGPG-GEGFMRLNAGCSLATLQDGLRRIKAALS------PatB

*Fig. 1.* Alignment of Lcs with Lcd, MalY, Hly, and PatB. The four residues previously reported to be invariant in aminotransferases (34, 35) are marked by closed circles. The  $Lys_{233}$  ( $K_{233}$ ) residue is the potential binding site of PLP. The eight residues that are found in most aminotransferases are boxed. Asterisks indicate amino acids that are identical in all five sequences. Lcs, Lcs protein from *Prevotella intermedia* (this study); Lcd, Lcd protein from *Streptococcus anginosus* (45); MalY, MalY protein from *Escherichia coli* (49); Hly, Hly protein from *Treponema denticola* (6); and PatB, PatB protein from *Bacillus subtilis* (1).

previously reported as L-cysteine desulfhydrase. The PI0305 gene on the genome of strain 17 was found as the homologue with high similarity. Based on the PI0305 gene sequence from strain 17, we determined the entire DNA sequence (1170 bp) of the PI0305 gene from P. intermedia ATCC 25611 and named this gene lcs. The lcs gene of strain ATCC25611 and the PI0305 gene from strain 17 shared 98% identity in the nucleotide sequence. The deduced amino acid sequence revealed that this *lcs* gene encoded a protein consisting of 390 amino acid residues with a calculated molecular mass of 44,381 Da and an isoelectric point of 5.61. Lcs showed 35% identity with Hly and Lcd, 39% with MalY, and 33% with PatB. As shown in Fig. 1, the aligned amino acid sequence of Lcs with Hly, Lcd, MalY, and PatB showed four invariant residues found in the  $\alpha$ -family of aminotransferases, and one of them was the putative PLP-binding residue, Lys (K), at position 233 in the Lcs (34, 35). In addition, six of the eight residues that are found in most aminotransferases in the  $\alpha$ -family are conserved.

# Purification of Lcs from *P. intermedia* ATCC25611

To confirm that the nucleotide sequence determined in this study specifies the Lcs enzyme, pCDS1 containing the *lcs* gene homologue was expressed in *E. coli* BL21 and purified by the GST purification system. The recombinant Lcs was purified as a single band with the molecular weight of 44 kDa (Fig. 2A), which was consistent with the predicted molecular mass

(44.4 kDa). The enzyme activity of the recombinant Lcs was confirmed by active staining on native-PAGE (see Assay II). As shown in Fig. 2C (lane 2), Lcs activity was detected as a single black band on the gel, confirming that the recombinant Lcs protein has the ability to produce hydrogen sulfide from L-cysteine. In addition, Lcs activity in crude extract from *P. intermedia* ATCC25611 was also detected as identical in size to the recombinant Lcs (Fig. 2C, lane 1) although Coomassie blue staining did not show an apparent Lcs protein band in the crude extract (Fig. 2B, lane 1).

# Characterization of the P. intermedia Lcs

To characterize the purified recombinant Lcs enzyme, we examined substrate specificity, kinetic parameters ( $K_{\rm m}$  and  $V_{\rm max}$ ), and optimum pH of the enzyme. The substrate specificity of the enzyme was examined by its ability to generate hydrogen sulfide from the SH-containing compounds such as L-cysteine, DL-homocysteine, cysteinyl glycine, and glutathione. The L-cysteine was a sole substrate for the P. intermedia Lcs enzyme, but DL-homocysteine, cysteinyl glycine, and glutathione were not substrates to produce hydrogen sulfide (data not shown). The  $K_{\rm m}$  and  $V_{\rm max}$  were determined for Lcs activity with L-cysteine as a substrate. The  $K_{\rm m}$  and  $V_{\rm max}$  were 0.7 mM and 4.2 µmol/min/mg of protein, respectively. The effect of pH on Lcs activity was tested by using 50 mM phosphate buffer at a pH range of 5.5-8.0. Maximum enzyme activity (hydrogen sulfide production) was observed at pH 7.0.

## Essential amino acids for Lcs activity

The crystal structure of Hly from T. denticola proposed the active site structure and the catalytic mechanism for Hly (28). suggesting that a part of the amino acid residues in the proposed active site are closely related to the catalytic reaction mechanism. On the basis of this report, we focused on amino acid residues such as Tyr-64, Tyr-123, Asp-203, Lys-238, and Arg-369, which are responsible for the reaction mechanism of Hly, and searched for the corresponding residues on the Lcs sequence. Therefore, the five amino acid residues (Tyr-59, Tyr-118, Asp-198, Lys-233, Arg-365) in the Lcs molecule were targeted and converted to Ala, Ala, Asn, Asn, and Gly, respectively, by site-directed mutagenesis, generating Tvr-59-Ala (Y59A), Tyr-118-Ala (Y118A), Asp-198-Asn (D198N), Lys-233-Asn (K233N), and Arg-365-Gly (R365G) mutants. As shown in Fig. 3A, each mutant was successfully expressed in E. coli BL21 and purified as a homogeneous protein with a single band of 44 kDa on sodium dodecyl sulfate-PAGE.

As shown in active staining (Fig. 3B), Y59A, Y118A, D198N, and K233N failed to display enzyme activity, implying that Tyr-59, Tyr-118, Asp-198, and Lys-233 were essential for Lcs enzyme activity. Lcs activity was also confirmed by hydrogen sulfide production using Assay I. Y59A, Y118A, D198N, and K233N did not exhibit hydrogen sulfide production although only R365G retained catalytic activity as well as the wild-type enzyme (data not shown).



*Fig.* 2. Polyacrylamide gel electrophoresis (PAGE) profile of L-cysteine desulfhydrase. (A) Sodium dodecyl sulfate–PAGE (12.5% acrylamide) of recombinant Lcs purified from *Escherichia coli* cells. The gel was stained with Coomassie brilliant blue. (B, C) Native PAGE (12.5% acrylamide) of Lcs enzyme. The sample was analysed by native PAGE and then the gel was stained with (B) Coomassie brilliant blue or (C) activity staining. Lanes 1, crude extract from *Prevotella intermedia* ATCC 25611; lanes 2, recombinant Lcs purified from *Escherichia coli* cells; M, size marker. Crude extract was prepared as follows: cell suspension of strain ATCC25611 was sonicated and centrifuged. The resultant supernatant was used as a crude extract.





## Effects of metal ions on Lcs activity

As it is reported that the zinc ion inhibits the formation of volatile sulfur compounds (24, 48, 49), we examined the effects of divalent metal ions containing the zinc ion on Lcs activity. Effects of divalent metal ions on Lcs activity were examined by the production of hydrogen sulfide (Assay I) and pyruvate (Assay III). CaCl<sub>2</sub>, MgCl<sub>2</sub>, MnCl<sub>2</sub>, or ZnCl<sub>2</sub> was added to the Lcs reaction mixture at a final concentration of 1 mM in each assay. Only ZnCl<sub>2</sub> strongly inhibited the production of hydrogen sulfide, but the other metal ions showed essentially no effect (data not shown). In addition, both pyruvate and hydrogen sulfide were measured under the presence of 0.01-1 mM ZnCl<sub>2</sub>. Production of both reaction end-products decreased gradually when increased amounts of ZnCl<sub>2</sub> were added to the Lcs reaction mixture (Table 1). In contrast, the addition of CaCl<sub>2</sub>, MgCl<sub>2</sub>, or MnCl<sub>2</sub> did not influence pyruvate production (data not shown).

# Hemolysis and hemoglobin-modification

It has been suggested that hydrogen sulfide modifies hemoglobin to primarily sulfhemoglobin and methemoglobin (29, 47), and the modified hemoglobin results in the lysis of RBCs (47). Therefore, we examined the effect of Lcs on hemolysis and hemoglobin-modification. As shown in Fig. 4A, RBCs lysed in the reaction mixture with both Lcs and L-cysteine although hemolysis was not observed with either Lcs or L-cysteine as a control. We also confirmed that incubation of RBCs with another reaction product such as pyruvate or ammonia did not produce hemolysis (data not shown). In this reaction, production of hydrogen sulfide reached a maximum at 30 min incubation (data not shown), which preceded hemolysis and hemoglobin-modification.

In addition, we measured the amounts of hemoglobin and modified-hemoglobin in the unbroken RBCs in the Lcs reaction mixture. As described in Materials and methods, lysate obtained by lysis of RBCs was used as the sample for determining hemoglobin (at  $OD_{540}$ ) and modified-hemoglobin (at  $OD_{620}$ ) (47). As shown in

Fig. 4, the amount of hemoglobin in the unbroken RBCs decreased in a timedependent manner for the initial 6-h incubation (Fig. 4B). In contrast, the amount of modified-hemoglobin increased in a timedependent manner (Fig. 4C). These findings suggest that hemoglobin changed to modified-types of hemoglobin in RBCs.

# Discussion

P. intermedia is thought to be closely associated with a variety of periodontal diseases and are able to produce various virulence factors that play an important role in the initiation and progression of periodontal diseases. Hemolytic activity contributes to the pathogenicity of P. intermedia (39, 43); however the mechanisms of hemolysis are unclear. In this study, we characterized the enzyme with hemolytic activity from P. intermedia ATCC25611. This enzyme catalysed L-cysteine to produce hydrogen sulfide, pyruvate, and ammonia, demonstrating that the enzyme is L-cysteine desulfhydrase. We named this enzyme Lcs. This type of enzyme has been genetically identified and characterized biochemically as Hly, Lcd, MalY, and PatB from T. denticola (9), S. anginosus (47), E. coli (50), and Bacillus subtilis (1), respectively. These reports reveal that these four enzymes belong to the  $\alpha$ -family of PLP-dependent enzymes. As shown in Fig. 1, alignment data showed that Lcs retained four invariant residues found in the  $\alpha$ -family of aminotransferases, and one (Lys) of which was the putative PLP-binding residue (34, 35). In addition, six of the eight residues that are found in most aminotransferases in the  $\alpha$ -family were conserved. These findings indicate that Lcs also belongs to the  $\alpha$ -family of aminotransferase and functions as a PLP-dependent cysteine desulfhydrase.

Substrate specificity of the Lcs enzyme revealed that only L-cycteine residue is a possible substrate. However, Lcs was inactive against the SH-containing di- and tri-peptides such as L-cysteinyl glycine and glutathione. It is known that periodontal pathogens use proteins for their nutrition and therefore produce a variety of proteolytic enzymes that are capable of

Table 1. Effect of ZnCl<sub>2</sub> on recombinant L-cysteine desulfhydrase

ZnCl <sub>2</sub>	Production (µmol/min/mg)			
	0 mM	0.01 mm	0.1 mm	1 mM
Pyruvate	5.26	4.76	2.53	< 0.01
Hydrogen sulfide	8.37	7.34	4.70	< 0.01

degrading periodontal tissues, tissue fluid, and immune proteins (14). As a result of the proteolytic degradation, proteins are ultimately broken-down into a variety of peptides and amino acids in the confines of the periodontal environment. L-Cysteine is likely to be present in this environment. Therefore, we speculate that the *P. intermedia* Lcs enzyme uses a cysteine which is supplied as free amino acid released from peptides by proteolytic degradation of periodontal pathogens including *P. intermedia*.

The crystal structure of cystalysin (Hly) from T. denticola proposed the active site structure and the catalytic mechanism for Hly (28). The superposition of the active sites of Hly with MalY revealed similar active site architectures and the conserved active site residues. In addition, the detailed catalytic mechanism for Hly suggested that part of the amino acid residues in the active site are closely responsible for catabolism of the substrate (28). Based on this information, we focused on the amino acid residues shown in the molecular reaction mechanism of Hly and chose five residues, Tyr-64, Tyr-123, Asp-203, Lys-238, and Arg-369, in the active site of Hly. Of these five residues, Tyr-64, Tyr-123, and Lys-238 are also suggested as the active site key residues in catalysis of a PLP-dependent enzyme reaction by Hly, MalY, and cystathionine  $\beta$ -lyase (28). Therefore, we searched the corresponding amino acid residues on the Lcs sequence and found five residues: Tyr-59, Tyr-118, Asp-198, Lys-233, and Arg-365. Replacement of these amino acids by site-directed mutagenesis exhibited that Tyr-59, Tyr-118, Asp-198, and Lys-233 are essential for Lcs enzyme activity (Fig. 3). This is the first report that identified the amino acids that are indispensable for the activity of L-cysteine desulfhydrase using a mutation study. Further mutational analysis will be needed to clarify other amino acid residues responsible for Lcs enzyme activity.

Zinc ions inhibit the formation of VSC, and thereby reduce or inhibit oral malodor (24, 48, 49). It is thought that zinc ions form insoluble sulfides by interacting with the sulfur in precursors of VSC, so inhibiting further production of VSC. In the present study, production of hydrogen sulfide in a Lcs reaction mixture was completely inhibited by 1 mM ZnCl<sub>2</sub> (Table 1). In this case, at least two inhibition modes could be suggested: one is formation of insoluble sulfides by interacting zinc ions with hydrogen sulfide as described above, and the other is inhibition



*Fig.* 4. Kinetics of hemolytic activity (A). Changes in the amounts of hemoglobin (B) and modifiedhemoglobin (C) in unbroken red blood cells (RBCs). The reaction mixture containing L-cysteine, L-cysteine desulfhydrase, and RBCs was incubated at 37°C with slow rotation for the time indicated. After centrifugation, the pellets (unbroken RBCs) were broken by the addition of distilled water to release hemoglobin (B) and modified-hemoglobins (C) from unbroken RBCs.  $\Box$ , RBCs only;  $\blacksquare$ , RBCs plus 1 mM L-cysteine;  $\blacktriangle$ , RBCs plus L-cysteine desulfhydrase;  $\blacklozenge$ , RBCs plus 1 mM L-cysteine plus L-cysteine desulfhydrase. Each graph is representative of three similar experiments.

of the Lcs enzyme by zinc ions. As shown in TABLE 1, production of not only hydrogen sulfide but also pyruvate gradually decreased in the presence of increased amounts of ZnCl<sub>2</sub>. Neither end-product was detected under the presence of 1 mM ZnCl<sub>2</sub>. This result strongly suggests that zinc ions interact with the Lcs enzyme (Lcysteine desulfhydrase), resulting in loss of Lcs enzyme activity. This also means that zinc ions have an excellent anti-hydrogen sulfide effect.

We suggested that the Lcs enzyme from *P. intermedia* is a cysteine-dependent enzyme that produces hydrogen sulfide, pyruvate, and ammonia as end-products and is responsible for hemoglobin-modification and hemolysis. Although these phenomena were observed only when RBCs were exposed to Lcs and L-cysteine (Fig. 4A,C), exposure of RBCs to either pyruvate or ammonia, or both, did not influence hemolysis and modification of

hemoglobin (data not shown). The result revealed that hydrogen sulfide produced by Lcs is closely associated with hemoglobin-modification and hemolysis. These observations are supported by the following reports. Moxness et al. (37) provided a general mechanism to connect sulfhemoglobin formation to erythrocyte lysis. A series of Hly studies have shown that hemoglobin modification by hydrogen sulfide always precedes hemolysis and that the cysteine-dependent lysis of erythrocytes caused by Hly is the result of hydrogen sulfide production, resulting in the accumulation primarily of sulfhemoglobin and methemoglobin (6-10). Our present finding also showed that, in the Lcs reaction mixture, the amount of hydrogen sulfide produced by the Lcs enzyme reached a maximum at about 30-min of incubation (data not shown); subsequently, lysis of RBCs and modification of hemoglobin were observed (Fig. 4). These results suggest that hydrogen sulfide modifies hemoglobin and the formation and accumulation of modifiedhemoglobin seem to play an important role in cysteine-dependent hemolysis as proposed by Hly studies (6–10).

Iron and heme are essential nutrients for periodontal pathogens and play a crucial role in their pathogenesis. Within the human host the majority of iron is found intracellularly in the form of hemoglobin. heme proteins, or ferritin (16) and small quantities of extracellular iron are complexed to iron-binding proteins, primarily transferrin and lactoferrin. Therefore, under physiological conditions, free hemoglobin or hemin may not be available because these compounds are efficiently scavenged by serum proteins haptoglobin and hemopexin, respectively (13, 16, 38). To survive in the iron-limited environment of the host, periodontal pathogens have multiple iron acquisition systems. P. intermedia, the etiological agent of periodontal diseases, also requires iron for growth (18), but lacks the ability to synthesize heme, an important growth factor for the bacterium. P. intermedia has the ability to bind to hemoglobin and utilize hemoglobin as an effective heme source (18, 31, 32) although this bacterium cannot utilize inorganic iron and iron-binding proteins such as transferrin and lactoferrin for growth. indicating that hemoglobin appears to be a more effective iron source (19, 30). To acquire hemoglobin, P. intermedia has hemolytic and hemagglutinating activities (4, 39, 42). A hemolytic substance (hemolysin) activated by reductants has been partially purified (43) and the gene encoding multiple hemolytic domains has been cloned in P. intermedia (3), but their mechanisms for hemolysis are still unclear because hemolytic substances have not been identified and characterized. As compared with our Lcs results, the property of the hemolytic substance and the nucleotide sequence encoding multiple hemolytic domains were distinct from those of our Lcs enzyme. These observations indicate that P. intermedia has several mechanisms for hemolysis, one of which is cysteine-dependent hemolysis by the Lcs enzyme as presented here. Our findings also suggest that inhibition of the Lcs enzyme by ZnCl<sub>2</sub> is effective for the prevention of oral malodor and periodontal diseases.

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